

*N*-(3-Chlorophenyl)-2,2,2-trimethylacetamideB. Thimme Gowda,<sup>a\*</sup> Sabine Foro<sup>b</sup> and Hartmut Fuess<sup>b</sup><sup>a</sup>Department of Chemistry, Mangalore University, Mangalagangotri 574 199, Mangalore, India, and <sup>b</sup>Institute of Materials Science, Darmstadt University of Technology, Petersenstrasse 23, D-64287 Darmstadt, Germany

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## Key indicators

Single-crystal X-ray study  
*T* = 299 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.015 \text{ \AA}$   
*R* factor = 0.071  
*wR* factor = 0.153  
Data-to-parameter ratio = 8.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

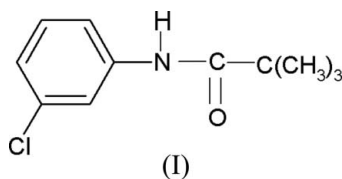
In the structure of the title compound (3CPTMA),  $\text{C}_{11}\text{H}_{14}\text{ClNO}$ , the conformation of the N—H bond is *anti* to the *meta*-chloro substituent, similar to that observed for the corresponding *meta*-methyl-substituted amide. The bond parameters of the two amides are also similar. The molecules in 3CPTMA are linked into chains, as bent columns, through N—H...O hydrogen bonding.

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## Comment

The structure of *N*-(3-chlorophenyl)-2,2,2-trimethylacetamide (3CPTMA), (I), has been determined as part of a study on the systematization of the crystal structures of *N*-aromatic amides (Gowda *et al.*, 2004; Gowda, Kozisek & Fuess, 2006; Gowda, Kozisek, Svoboda & Fuess, 2007). In 3CPTMA, the conformation of the N—H bond is *anti* to the *meta*-chloro substituent (Fig. 1), similar to that observed for 2,2,2-trimethyl-*N*-(3-methylphenyl)acetamide (Gowda, Kozisek, Tokarcik & Fuess, 2007). The bond parameters of the two amides and those of the ring-unsubstituted 2,2,2-trimethyl-*N*-phenylacetamide are similar (Gowda, Kozisek, Tokarcik & Fuess, 2007). The molecules in 3CPTMA are linked into chains, as bent columns, through N—H...O hydrogen bonding (Table 1), similar to that observed in other amides (Gowda, Paulus *et al.*, 2007; Gowda, Kozisek, Tokarcik & Fuess, 2007).



## Experimental

The title compound was prepared according to the literature method (Gowda, Shilpa *et al.*, 2006). The purity of the compound was checked by determining its melting point. It was characterized by recording its IR and NMR spectra (Gowda, Shilpa *et al.*, 2006). Single crystals of the title compound were obtained by slow evaporation of an ethanol solution and used for X-ray diffraction studies at room temperature.

## Crystal data

$\text{C}_{11}\text{H}_{14}\text{ClNO}$	$V = 1158.3 (3) \text{ \AA}^3$
$M_r = 211.68$	$Z = 4$
Orthorhombic, $Pca2_1$	Cu $K\alpha$ radiation
$a = 10.778 (1) \text{ \AA}$	$\mu = 2.66 \text{ mm}^{-1}$
$b = 10.623 (1) \text{ \AA}$	$T = 299 (2) \text{ K}$
$c = 10.117 (2) \text{ \AA}$	$0.68 \times 0.05 \times 0.03 \text{ mm}$

Data collection

Enraf–Nonius CAD-4 diffractometer  
Absorption correction:  $\psi$  scan (North *et al.*, 1968)  
 $T_{\min} = 0.754$ ,  $T_{\max} = 0.855$  (expected range = 0.814–0.923)  
1745 measured reflections

1095 independent reflections  
479 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.152$   
3 standard reflections  
frequency: 120 min  
intensity decay: 5%

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.071$   
 $wR(F^2) = 0.153$   
 $S = 0.99$   
1095 reflections  
127 parameters  
1 restraints

H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.23 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.28 \text{ e \AA}^{-3}$   
Absolute structure: (Flack, 1983)  
Flack parameter: 0.01 (7)

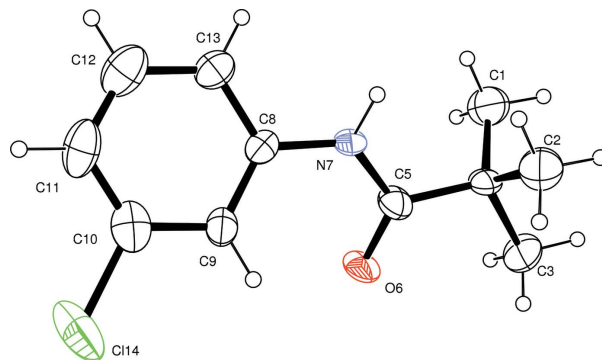


Figure 1

The molecular structure of the title compound, showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are represented as small spheres of arbitrary radius.

Table 1

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N7-H7N\cdots O6^i$	0.86	2.10	2.935 (8)	164

Symmetry code: (i)  $-x - \frac{1}{2}, y, z - \frac{1}{2}$ .

All H atoms were positioned geometrically and treated as riding, with  $C-H = 0.93$  (CH aromatic) or  $0.96 \text{ \AA}$  ( $\text{CH}_3$ ) and  $N-H = 0.86 \text{ \AA}$ , with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{NH})$  and  $1.5U_{\text{eq}}(\text{methyl C})$ .

Data collection: *CAD-4-PC* (Nonius, 1996); cell refinement: *CAD-4-PC*; data reduction: *REDU4* (Stoe & Cie, 1987); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia 1997); software used to prepare material for publication: *SHELXL97*.

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